

①

## Lecture 36

### Heisenberg's uncertainty principle

- It is not possible to assign determinate values to both the position & momentum of a particle
- recall that the state of a quantum particle is not described by a list of numbers but instead a wavefunction from which we can compute probabilities

(2)

In quantitative terms, uncertainty principle is

$$\Delta x \cdot \Delta p_x \geq \hbar = \frac{h}{2\pi} \quad \begin{matrix} \downarrow \\ \text{Planck} \end{matrix} \quad \begin{matrix} \downarrow \\ \text{constant} \end{matrix}$$

↑  
uncertainty

(standard deviation)

when measuring

x component of

position

↑  
uncertainty  
(standard deviation)

when measuring

x component  
of

momentum

~~lower bound is small but~~  
it matters when dealing  
w/ effects @ quantum  
scale.

(3)

To interpret, suppose  
that we can make uncertainty  
of position  $x$  very small.

Then necessarily uncertainty  
of momentum  $p_x$  must be large.

There is also uncertainty principle  
for  $y + z$  coordinates:

$$\Delta y \cdot \Delta p_y \geq \hbar$$

$$\Delta z \cdot \Delta p_z \geq \hbar$$

We won't derive them here

b/c we need more background  
in quantum mechanics  
to do that!

④

Q: Suppose that an electron is moving along an  $x$  axis and that you measure its speed to be  $2.05 \times 10^6 \text{ m/s}$  with precision of  $\pm 5\%$ .

What is minimum uncertainty w/ which you can measure its position?

Note that  $\Delta p_x = (0.005) p_x$ .

A:

$$p_x = m \cdot v_x = (9.11 \times 10^{-31} \text{ kg})$$

$$(2.05 \times 10^6 \text{ m/s})$$

$$= 1.87 \times 10^{-24} \frac{\text{kg} \cdot \text{m}}{\text{s}}$$

$$\Rightarrow \Delta p_x = 9.35 \times 10^{-27} \frac{\text{kg} \cdot \text{m}}{\text{s}}$$

$$\Rightarrow \Delta x = \frac{\hbar}{\Delta p_x} = \frac{6.63 \times 10^{-34} \text{ J} \cdot \text{s}}{9.35 \times 10^{-27} \text{ kg} \cdot \text{m/s}}$$

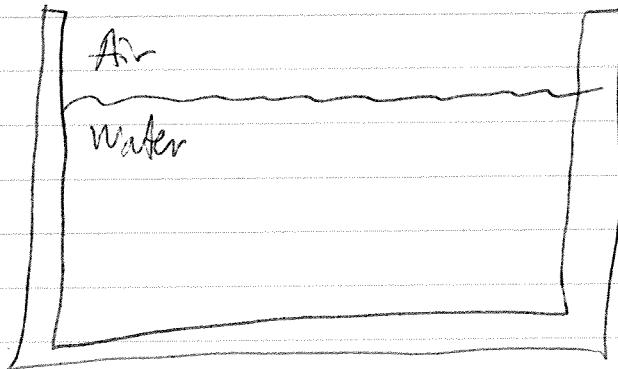
(5)

$$= 11 \text{ nm}$$

( $\approx$  100 atomic diameters)

## 14-2 Fluids at rest

Consider a tank of water



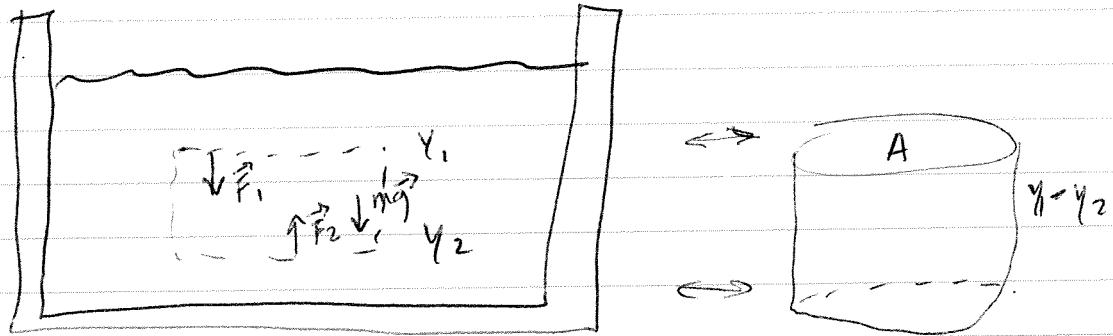
- pressure increases w/ depth below air-water interface.
- we are interested in hydrostatic pressure, due to fluid @ rest, as a function of depth.

(6)

How to analyze? Consider forces

acting on a cylindrical region in

tank



- downward force  $\vec{F}_1$  due to water above cylinder
- upward force  $\vec{F}_2$  due to water below cylinder
- ~~$\vec{mg}$~~  is downward gravitational force due to mass of water in cylinder.

7

balance of forces B (sum of forces = 0  
for static  
scenarios)

$$F_2 = F_1 + mg$$

In terms of pressures, this is

$$F_1 = p_1 A \quad , \quad F_2 = p_2 A$$

mass of water in cylinder B

$$m = \rho V$$

$\uparrow$   $\uparrow$   
density of water volume of water

so mass of water

$$m = \rho A \underbrace{(y_1 - y_2)}$$

volume of cylinder

$\Rightarrow$

$$p_2 A = p_1 A + \rho A (y_1 - y_2) g$$

$$\Rightarrow p_2 = p_1 + \rho (y_1 - y_2) g$$

(8)

can use this to find pressure  
below surface.

can also use it to find pressure  
in atmosphere.

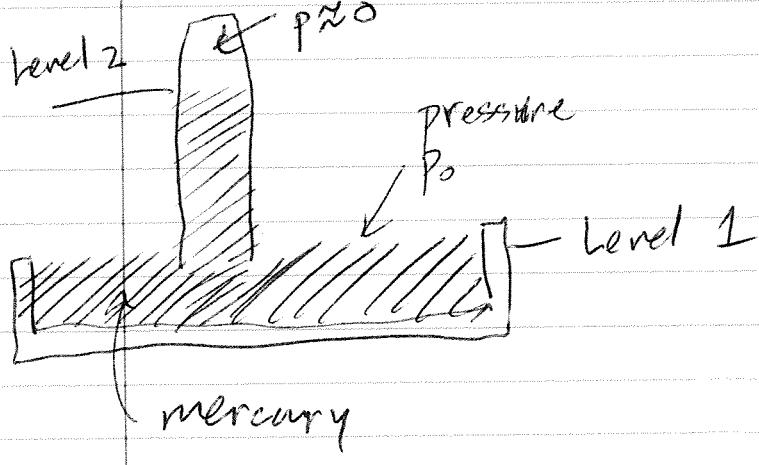
pressure @ a point in a static

fluid depends only on depth

& not on horizontal dimension.

### Measuring Pressure

#### Mercury Barometer



Can use previous  
equation to calculate  
atmospheric pressure  $p_0$   
in terms of height  
 $h$  of mercury  
column.

⑨

$$\gamma_1 = 0, P_1 = P_0$$

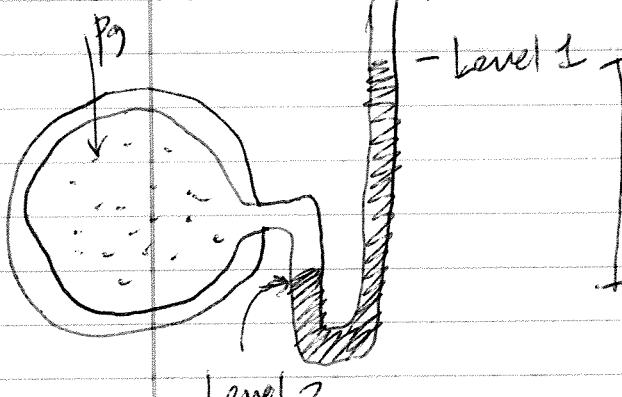
$$\gamma_2 = h, P_2 = 0$$

$$\Rightarrow P_0 = \rho gh$$

where  $\rho$  is density of mercury.

Another way to measure

open-tube manometer measures  
gauge pressure of a gas



called a  
"U-tube"

can again use  
equation to measure

pressure  $P_g$ .

(10)

$$y_1 = 0 \quad p_1 = p_0$$

$$y_2 = -h \quad p_2 = p$$

$$\Rightarrow p_g = p - p_0 = \rho g h$$

$\rho$   
density of liquid

Gauge pressure can be positive  
or negative (depending on  
whether  $p > p_0$  or  $p < p_0$ )